

Lab: Dynamic Equilibrium Simulation

FOR THE TEACHER

Summary

In this lab, students will explore equilibrium using paper clips to mimic a chemical reaction.

Resource Type
Lab

Grade Level
High School

Objectives

By the end of this lesson, students should be able to

- Better understand what it means for a system to reach equilibrium.

Chemistry Topics

This lesson supports students' understanding of

- Equilibrium
- Le Châtelier's Principle

Time

Teacher Preparation: 30 minutes

Lesson: 1 hour

Materials

For each group:

- ~60 paper clips
- timer

Safety

No notable safety precautions need to be observed for this lab.

Teacher Notes

- Students should work in groups of four: Two people will put paper clips together, one person will take paper clips apart, and one person will be the timer and record data.
- Students are required to generate their own data table for the three parts.

FOR THE STUDENT

Student Activity Sheet: Dynamic Equilibrium Simulation Lab

Lesson

Background

You will be modeling the reaction:



by putting paper clips together (forward reaction) and taking them apart (reverse reaction). This simulation allows you to "stop" the reaction after a certain amount of time so that you can monitor the "concentrations" of the reactants and products.

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Thanks to:

Ward's Science

Dynamic Equilibrium Lab Manual

Terry C. Jones



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