

(ii) Calculate the percentage yield if 2.7 g of copper(II) sulfate-5-water is obtained from 0.020 mol of sulfuric acid.

(2)

(iii) What is the most likely reason for the yield being well below 100%?

(1)

(d) When the crystals are heated, they turn white. On adding water, they return to their original colour. Suggest a use for this reaction.

(1)

(Total for Question 20 = 15 marks)



P 4 1 6 4 9 A 0 1 5 2 4

Edexcel Chemistry Unit 1 June 2013 Paper

Lauren Gardner



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Chemistry Scottish Consultative Council on the Curriculum. Higher Still Development Unit, 2000
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Practical Chemistry

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