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Semiconductor Physics and Devices: Basic Principles, 4th edition By D. A. Neamen

Chapter I Problem Solutions

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Chapter 1

Problem Solutions

1.1

- (a) foc: 8 corner asoms ×1/8 = 1 asom 6 face atoms ×1/2 = 3 atoms Total of 4 atoms per unit cell
- (b) boc: 8 corner atoms ×1/8 = 1 atom. I enclosed atom -1 atom. Total of 2 atoms per unit cell
- (c) Diamond: 8 corner atoms ×1/8 = 1 atom 6 face atoms ×1/2 = 3 atoms 4 enclosed atoms = 4 atoms Total of 8 atoms per unit cell

1.2

(a) Simple cubic faction: a = 2x Unit cell vol = $a^3 = (2r)^2 = 8r^3$ 1 atom per cell, so atom vol = $\left(1\right)\frac{4\pi r^2}{4}$

Theorem.

Ratio =
$$\frac{\left(\frac{4\pi r^3}{3}\right)}{8r^3} \times 100\% = 52.4\%$$

(b) Pace-centered cubic lattice $d = 4r = a\sqrt{2} \Rightarrow a = \frac{d}{\sqrt{2}} = 2\sqrt{2} \cdot r$

Unit cell vol =
$$a^3 = \left(2\sqrt{2} \cdot r\right)^3 = 16\sqrt{2} \cdot r^3$$

4 atoms per cell, so atom vol = $\left(4\right)\left(\frac{4\pi r^3}{3}\right)$

Theory.

Ratio =
$$\frac{\left(4\left(\frac{4.8 \, r^3}{3}\right)}{16\sqrt{2} \cdot r^2} \times 100\% = 74\%$$

(c) Body-centered cubic lattice

$$a = 4r = a\sqrt{3} \Rightarrow a = \frac{4}{\sqrt{3}} \cdot r$$

Unit cell vol =
$$a^3 = \left(\frac{4}{\sqrt{3}} \cdot r\right)^3$$

2 atoms per cell, so atom vol =
$$(2)\left(\frac{4\pi x^3}{3}\right)$$

Ti become

Ratio =
$$\frac{\left(2\left(\frac{4\pi r^3}{3}\right)}{\left(\frac{4r}{\sqrt{3}}\right)^3} \times 100\% = 63\%$$
Diamond lattice

(d) Diamond lattice

Body diagonal =
$$a = 8r = a\sqrt{3} \Rightarrow a = \frac{8}{\sqrt{3}}$$

Unit cell vol =
$$a^3 = \left(\frac{8r}{\sqrt{3}}\right)^3$$

8 atoms per cell, so atom vol = (8)
$$\left(\frac{4\pi r^3}{3}\right)$$

Theory

Ratio =
$$\frac{(8(\frac{4.9 \text{ r}^3}{3})}{(\frac{3 \text{ r}}{\sqrt{3}})^3} \times 100\% = 34\%$$

1.3

(a) α = 5.43 A; From Problem 1.2d.

Then
$$r = \frac{a\sqrt{3}}{8} = \frac{(5.43)\sqrt{3}}{8} = 1.176 \text{ Å}$$

Center of one silicon atom to center of nearest neighbor = 2r = 2.35 A

(b) Number density

$$= \frac{8}{(5.43 \times 10^{-6})^3} = 5 \times 10^{21} \text{ cm}^{-6}$$

(c) Mass demotry

$$= \rho = \frac{N(AtWt)}{N_{*}} = \frac{(5 \times 10^{32})(28.09)}{6.02 \times 10^{32}}$$

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