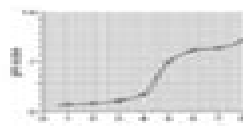


4. State the colour of universal indicator in solutions of:
- an acid \_\_\_\_\_ (1 mark)
  - an alkali \_\_\_\_\_ (1 mark)
5. a Do acids taste sweet or sour? \_\_\_\_\_ (1 mark)
- b Which is dangerous, acid, alkali or they can both be dangerous? \_\_\_\_\_ (1 mark)

6. AaliF wanted to investigate the pH changes when an acid reacts with an alkali.

He put 5 cm<sup>3</sup> of the acid in a tube.  
He added the alkali 1 cm<sup>3</sup> at a time.  
Each time he added the alkali, he measured the pH of the solution.  
His results are shown on the graph.



- a. Suggest a way in which AaliF could have measured the pH of the solution. \_\_\_\_\_ (1 mark)
- b. State the volume of alkali AaliF needed to make the solution neutral. \_\_\_\_\_ (1 mark)
- c. Explain what the graph shows. \_\_\_\_\_ (2 marks)

7. Answer the following questions using one of the substances in the box below.
- |            |              |         |       |              |
|------------|--------------|---------|-------|--------------|
| pure water | citric juice | calcium | sugar | oven cleaner |
|------------|--------------|---------|-------|--------------|
- Which substance contains an acid? \_\_\_\_\_ (1 mark)
  - Which substance contains an alkali? \_\_\_\_\_ (1 mark)
  - Which substance changes colour to show an acid or an alkali? \_\_\_\_\_ (1 mark)
8. State what each of these hazard symbols means.



- a. \_\_\_\_\_ (1 mark)
- b. \_\_\_\_\_ (1 mark)

2. A bottle of acid is labelled with this hazard symbol.
- 
- State two safety precautions you would take before using that acid in an experiment.
- Precaution 1: \_\_\_\_\_ (1 mark)

7. A substance has a pH of 12.
- State the hazard symbol that should be on a bottle. \_\_\_\_\_ (1 mark)
- Is the substance an acid or an alkali? \_\_\_\_\_ (1 mark)

8. An indicator called bromothymol blue is yellow in an acid and blue in an alkali.  
Suggest the colour of bromothymol blue in a neutral solution. \_\_\_\_\_ (1 mark)

9. The word equation for a reaction is:  
nitric acid + potassium hydroxide → potassium nitrate + water
- State the names of the two reactants in the word equation. \_\_\_\_\_ (2 marks)
  - State the name of the salt in the word equation. \_\_\_\_\_ (1 mark)

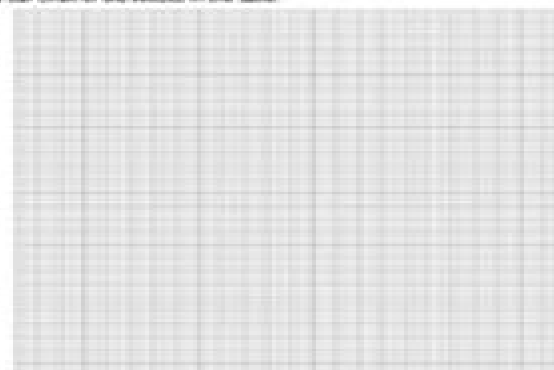
10. Hydrochloric acid reacts with calcium oxide to make calcium chloride and water.  
Write the word equation for this reaction. \_\_\_\_\_ (2 marks)

11. Elen investigated the temperature rise when different acids were added to sodium hydroxide solution.

She measured 10 cm<sup>3</sup> of sodium hydroxide solution, and noted the temperature.  
She added 10 cm<sup>3</sup> of an acid to the sodium hydroxide solution and noted the new temperature.  
She calculated the temperature rise.  
She repeated the experiment with the same volume of the sodium hydroxide solution but added different acids to it.  
Her results are in the table.

Acid added to sodium hydroxide solution	Temperature rise (°C)
citric acid	3.5
hydrochloric acid	13.8
nitric acid	13.8
sulphuric	8.8

- State which variable has been changed in this experiment (the independent variable). \_\_\_\_\_ (1 mark)
- State which variable has been measured (the dependent variable). \_\_\_\_\_ (1 mark)
- State the word that describes the reaction between an acid and an alkali. \_\_\_\_\_ (1 mark)
- Put a bar chart of the results in the table. \_\_\_\_\_ (1 mark)



- Write a title for the chart. \_\_\_\_\_ (1 mark)

# Exploring Science Test

**Teen-Hang Meen, Stephen Prior, Artde  
Lam**



## **Exploring Science Test:**

*Discovering Science Through Inquiry: Earth Systems and Cycles Kit* Kathleen Kopp, 2010-07-14 The Discovering Science through Inquiry series provides teachers and students of grades 3-8 with direction for hands-on science exploration around particular science topics and focuses. The series follows the 5E model: engage, explore, explain, elaborate, evaluate. The Earth Systems and Cycles kit provides a complete inquiry model to explore Earth's various systems and cycles through supported investigation. Guide students as they make cookies to examine how the rock cycle uses heat to form rocks. Earth Systems and Cycles kit includes 16 Inquiry Cards in print and digital formats, Teacher's Guide, Inquiry Handbook. Each kit includes a single copy; additional copies can be ordered. Digital resources include PDFs of activities and additional teacher resources including images and assessment tools, leveled background pages for students, and video clips to support both students and teachers.

**Survey of Science History & Concepts Parent Lesson Plan**, 2013-08-01 Survey of Science History Concepts Course Description: Students will study four areas of science: Scientific Mathematics, Physics, Biology, and Chemistry. Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation. Each content area is thoroughly explored, giving students a good foundation in each discipline.

**Semester 1: Math and Physics** Numbers surround us. Just try to make it through a day without using any. It's impossible: telephone numbers, calendars, volume settings, shoe sizes, speed limits, weights, street numbers, microwave timers, TV channels, and the list goes on and on. The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them. It's amazing how ten simple digits can be used in an endless number of ways to benefit man. The development of these ten digits and their many uses is the fascinating story in *Exploring the World of Mathematics*. Physics is a branch of science that many people consider to be too complicated to understand. John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend. Did you know that a feather and a lump of lead will fall at the same rate in a vacuum? Learn about the history of physics from Aristotle to Galileo to Isaac Newton to the latest advances. Discover how the laws of motion and gravity affect everything from the normal activities of everyday life to launching rockets into space. Learn about the effects of inertia first hand during fun and informative experiments. *Exploring the World of Physics* is a great tool for student who want to have a deeper understanding of the important and interesting ways that physics affects our lives.

**Semester 2: Biology and Chemistry** The field of biology focuses on living things from the smallest microscopic protozoa to the largest mammal. In this book you will read and explore the life of plants, insects, spiders, and other arachnids, life in water, reptiles, birds, and mammals, highlighting God's amazing creation. You will learn about biological classification, how seeds spread around the world, long-term storage of energy, how biologists learned how the stomach digested food, the plant that gave George de Mestral the idea of Velcro, and so much more. For most of history, biologists used the visible appearance of plants or animals to classify them. They grouped

plants or animals with similar looking features into families Starting in the 1990 s biologists have extracted DNA and RNA from cells as a guide to how plants or animals should be grouped Like visual structures these reveal the underlying design of creation Exploring the World of Biology is a fascinating look at life from the smallest proteins and spores to the complex life systems of humans and animals Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information

**Innovation in Design, Communication and Engineering** Teen-Hang Meen,Stephen Prior,Artde Lam,2015-07-23 This volume represents the proceedings of the 2014 3rd International Conference on Innovation Communication and Engineering ICICE 2014 This conference was held in Guiyang Guizhou P R China October 17 22 2014 The conference provided a unified communication platform for researchers in a wide range of fields from information technology

**Science and Religion in Education** Berry Billingsley,Keith Chappell,Michael J. Reiss,2019-11-06 This book brings together the latest research in education in relation to science and religion Leading international scholars and practitioners provide vital insights into the underlying debates and present a range of practical approaches for teaching Key themes include the origin of the universe the theory of evolution the nature of the human person the nature of science and Artificial Intelligence These are explored in a range of international contexts The book provides a valuable resource for teachers students and researchers in the fields of education science religious education and the growing specialist field of science and religion Science and Religion in Education is a compelling read for current and future generations of academic researchers and teachers who wish to explore the fascinating intersect between science education and religious studies The research findings and insights presented by these international scholars offer new dimensions on contemporary practice

Vaile Dawson Professor of Science Education University of Western Australia Science and Religion in Education offers a fascinating and diverse collection of chapters surveying the current state of thinking about how science and religion can be understood in education The book offers a wealth of thought provoking material for anyone interested in the natures of science and religion their relationship s or their representation within the curriculum Professor Keith Taber University of Cambridge Science education and religious education are uncomfortable bedfellows This book written in part as a response to the perhaps too clear accounts of Ian Barbour provides suitably nuanced pictures of how science and religion are dealt with in schools Whatever the views of specialists young people receive an education in both science and religion hearing

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