

## Teacher Guide: Boyle's Law and Charles' Law



### Learning Objectives

- Observe the effects of pressure changes on a fixed volume of gas.
- Define pressure.
- Calculate the pressure volume the product of a gas at a constant temperature (Boyle's Law).
- Calculate the temperature change on the volume of a gas under constant pressure (Charles' Law).



### Materials

Students will require a few "Boyle's Law" and "Charles' Law" apparatus.



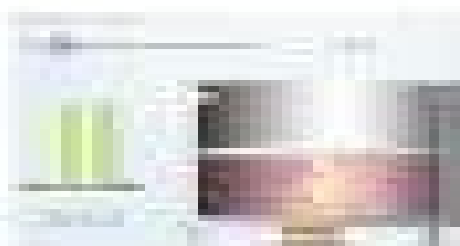
### Background

Boyle's Law states that the product of the pressure and volume of a gas is constant at a constant temperature. The product of the pressure and volume of a gas is constant at a constant temperature.

In the Boyle's Law experiment, a gas (air) is contained in a cylinder. The pressure of the gas is measured by a manometer. The volume of the gas is measured by a graduated cylinder.

The following procedure describes the experiment:

1. Boyle's Law: Students determine the relationship between pressure and volume of a gas at a constant temperature (Boyle's Law).
2. Charles' Law: Students determine the relationship between volume and temperature of a gas at a constant pressure (Charles' Law).



### Suggested Student Response

1. The student will observe the effect of pressure changes on a fixed volume of gas.

Students will observe a decrease in volume of a gas as the pressure increases. This is because the gas molecules are being pushed closer together as the pressure increases. The volume of the gas decreases as the pressure increases.

Students will observe a decrease in volume of a gas as the temperature increases. This is because the gas molecules are moving faster as the temperature increases. The volume of the gas decreases as the temperature increases.



# Explore Learning Boyles Law Teacher Guide

**Science Teacher Education  
Project, John Trevor Haysom, Clive  
Remer Sutton**



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**Prentice Hall Chemistry**, 2000      **Parent/Teacher Handbook** Edward A. Buchanan, 2004 This book will give you the guidance you need to share your own love of the Christian heritage with the children in your life

**Canadian Failures** Alex Benay, 2017-10-21 Ten Canadians make one powerful argument we cannot shy away from failure if we hope to succeed Canadian Failures gathers experts at the top of their field all of whom have grappled with failure including astronaut Robert Thirsk Olympic gold medalist wrestler Erica Wiebe and Tom Jenkins of OpenText Corporation

The Science Teacher, 2009      **Engineering Ingenuity** NAGC, 2025-12-30 Engineering Ingenuity Instruction for High Ability Learners in K 8 Classrooms offers a groundbreaking exploration of innovative strategies interdisciplinary approaches and cutting edge technologies designed to empower gifted and talented students in STEM education This updated edition dives deeper into the intersection of engineering creativity and advanced cognitive abilities providing educators with actionable frameworks to inspire the next generation of problem solvers and innovators Chapters explore interdisciplinary learning by integrating science mathematics reading and engineering design to create dynamic inquiry based experiences that foster critical thinking and innovation This new edition also highlights the transformative potential of artificial intelligence and machine learning in personalized learning addresses equity and inclusion in STEM by providing strategies to overcome barriers faced by underrepresented Black and Hispanic students and emphasizes the importance of guided play in developing early math and spatial skills Additionally it explores problem based learning to deeply engage students with engineering concepts and enrichment opportunities that nurture their interests and talents

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